



Introduction

Inorganic carbon parameters are commonly measured as part of oceanic carbon cycle studies. The four measurable CO₂ parameters are total CO₂ (TCO₂), total alkalinity (TA), total hydrogen ion concentration (pH_T) and partial pressure of CO₂ (pCO₂). All four CO₂ parameters can be measured on ships with excellent accuracy and precision. The dissolved inorganic carbon equilibria are constrained by measurement of two of these parameters. For example, if TCO₂ and pH_T are known, pCO₂, TA, CO₂(aq), H₂CO₃, HCO₃⁻, CO₃²⁻ and CaCO₃ saturation can be calculated. The objective of most studies is to quantify TA and TCO₂. Air-sea CO₂ flux calculations also require sea surface pCO₂ (e.g. Takahashi et al. 1997). Because pCO₂ is highly variable and not well-understood, it is often measured in carbon process studies (e.g. Oudot 1989; Robertson et al. 1993; Boyd et al. 2000). Depth-resolved TCO₂ measurements are used to quantify inorganic carbon budgets in the water column. Time-series of TCO₂ in the euphotic zone give important insights into the biological and physical mechanisms that control carbon fluxes (e.g. Chipman et al. 1993; Marchal et al. 1996; Gruber et al. 1999). Full water column TCO₂ inventories are being used to quantify the uptake of anthropogenic CO₂ (Gruber et al. 1996; Peng et al. 1998; Körtzinger et al. 1998).

A major initiative of oceanographic programs such as GOOS is to develop these measurements for autonomous characterization of the ocean carbon cycle. At present, only autonomous pCO₂ systems are fairly well-advanced (e.g. Friederich et al. 1995; Hood et al. 1999; DeGrandpre et al. 1995). Less progress has been made on the other inorganic carbon parameters. TCO₂ is probably the most technically challenging measurement because of the need for acidification and quantification of the liberated CO₂. Alternatively, TCO₂ can be calculated from measurements of pH_T and TA, or pCO₂ and TA (the pH-pCO₂ combination is prone to large TCO₂ errors, see Millero et al. 1993).

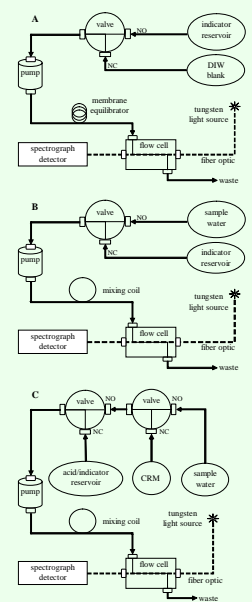
Figure 1: The SAMI-CO₂ in its mooring cage. The SAMI is a membrane-based sensor that measures the partial pressure of CO₂. It can operate for up to 1 year with a ~24/day measurement frequency. The instrument was developed for mooring applications but could easily be made more compact for smaller payload platforms.



Our past research has focused on the development of a CO₂ sensor named the submersible autonomous moored instrument for CO₂ (SAMI-CO₂) (DeGrandpre et al. 1995) (Figure 1). These sensors are now commercially available from Sunburst Sensors, LLC in Missoula, Montana. One of our present goals is to adapt this technology for pH_T and TA measurements for full characterization of inorganic carbon (Figure 2). The new sensors are named SAMI-pH and SAMI-alk because their designs derive directly from SAMI-CO₂.

Methods

Figure 1: Schematic diagrams of (A) SAMI-CO₂, (B) SAMI-pH and (C) SAMI-alk. Because all three analytical systems are based on spectrophotometric pH measurements, the designs use the same pump and valve configuration and optical detection schemes. SAMI-CO₂ does not actually sample the ambient seawater but uses a membrane to equilibrate with seawater pCO₂ (A). To perform pH_T measurements (B), the membrane is replaced with a mixing coil. The indicator is relocated to the normally-closed (NC) port of the 3-way valve and seawater is drawn through the normally-open (NO) port. The pH_T configuration is adapted for total alkalinity measurements by simply replacing the pure indicator solution with a HCl/indicator mixture. An additional 3-way valve is used to introduce an Alkalinity Certified Reference Material (Dickson et al. 2003).

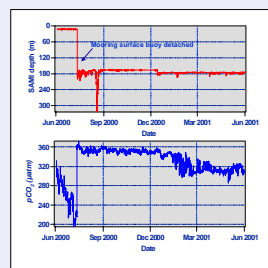


State of the Technology

SAMI-CO₂ (Figure 2A)

The SAMI-CO₂ is a well understood and proven technology that has been widely used in many marine field studies. These studies have been undertaken by our group and by a number of international collaborators. Deployments have been made on drifters and moorings in all of the world's oceans. SAMIs are currently deployed in the Beaufort Sea, the eastern North Atlantic (PAP time-series), western North Atlantic (CIS time-series), and eastern North Pacific (PAPA time-series). Results from a 1 year deployment in the Labrador Sea are shown in Figure 3.

Figure 3: 12 months of continuous SAMI-CO₂ data collected in the Labrador Sea (the surface flotation parted and the mooring sank to 180 m depth). Excellent long-term stability is apparent during the ~6 month period prior to winter deep convection (late January). However, no shipboard pCO₂ data were available at 180 m depth for validation of the pCO₂ data. During the latter half of the time-series, the SAMI-CO₂ was within the pycnocline creating a strong correlation between pCO₂ and temperature (not shown) (data were collected in collaboration with D. Wallace and U. Send, IfM, Kiel, Germany).



SAMI-pH (Figure 2B)

A prototype SAMI-pH instrument has been developed and tested for freshwater applications (Martz et al. 2003). The instrument is essentially identical in appearance to SAMI-CO₂ (Figure 1). An 8-day time-series is presented in Figure 4.

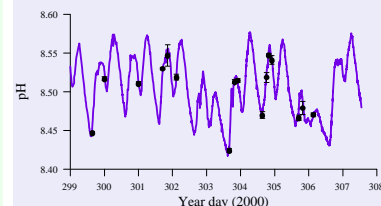


Figure 4: In situ SAMI-pH measurements (purple line) in the Clark Fork River. Lab spectrophotometric measurements are also shown (black circles).

SAMI-pH can be readily adapted for seawater pH measurements. Autonomous pH_T may have significant advantages over pCO₂ for inorganic carbon characterization on Lagrangian platform including a faster response time and potentially better reliability because pre-deployment gas phase (CO₂) calibrations are not required.

SAMI-alk (Figure 2C)

SAMI-alk is the least well-advanced of the three systems. We have recently obtained preliminary data demonstrating the feasibility of using the design in Figure 2C and a theoretical model has been developed to predict the sensor response. We are currently seeking grant support from NSF for continuation of this work.

Summary

A sensor combination that measures pCO₂ or pH_T with total alkalinity can be used to quantify TCO₂ and thereby provide very valuable information for understanding the mechanisms that control CO₂ exchange between the oceans and atmosphere. Developing this combination of sensors with sufficient precision and accuracy is feasible but will require a diligent, single-minded, bull-headed, long-term effort with an equal long-term commitment and patience by funding agencies. After all, conductivity sensors weren't developed overnight!

More background information:

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- DeGrandpre, M.D., Hammar, T.R. and C.D. Wirick. 1998. Short-term pCO₂ and O₂ dynamics in California coastal waters. *Deep-Sea Res.* II, 45, 1557-1575.
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Acknowledgments

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